Superelastic Porous NiTi with Adjustable Porosities Synthesized by Powder Metallurgical Method

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(Submitted February 14, 2012; in revised form June 14, 2012)

The effects of cold compaction pressure, TiH_2 , and NH_4HCO_3 on the physical and mechanical properties of porous NiTi fabricated by capsule-free-hot isostatic pressing (CF-HIP) were investigated. The porosities vary from 20 to 48% depending on the pressure. Porous NiTi with as great as 42% porosity and irregular pores can be prepared by the addition of TiH_2 , whereas 48% porosity and circular pores can be achieved by NH_4HCO_3 addition. All the porous NiTi fabricated by CF-HIP using these three methods exhibit super-elastic behavior that closely matches human bones.

Keywords implants, porous NiTi, powder metallurgy, superelastic

1. Introduction

Stainless steels and Ti-based alloys are used in the medical fields because of their high strength, good wear resistance, high corrosion resistance, and excellent biocompatibility (Ref 1-4). However, the high Young's moduli of these biometals impair bone regrowth when implanted in humans (Ref 4). Porous NiTi has recently become one of the most promising materials in surgical implants because of its open structure, low elastic modulus, high strength, good biocompatibility, and excellent superelasticity (SE) (Ref 5-10). The manufacturing processes and characteristics of porous NiTi have been reported (Ref 6, 10-18). For instance, Li et al. (Ref 11) and Chu et al. (Ref 12) have produced porous NiTi by combustion synthesis (CS). Unfortunately, the stress-strain curve obtained from their materials does not show an obvious stress plateau and the materials are brittle. Porous

This article is an invited paper selected from presentations at the International Conference on Shape Memory and Superelastic Technologies 2011, held November 6-9, 2011, in Hong Kong, China, and has been expanded from the original presentation.

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NiTi prepared by Li, Zhu, et al. employing elemental powder sintering (EPS) (Ref 13-15) exhibit low compression strength and low strain recovery. Kim et al. (Ref 16) have fabricated porous NiTi by self-propagating high temperature synthesis (SHS) and their materials also have relatively poor mechanical properties. The porous NiTi made by HIP by Lagoudas and Vandygriff (Ref 17) also shows low compression strength because of the irregular pore structure. Yuan et al. (Ref 18) have prepared recently a porous NiTi by CF-HIP that is used to improve the mechanical properties of ceramic materials while retaining adequate porosity of the as-pressed powders (Ref 19), but it is difficult to control the pore size, shape and distribution. To build up on our recent success (Ref 20), we report herein the production of porous NiTi by the CF-HIP with adjustable porosities, pore shape and size as well as good mechanical properties and excellent SE. We investigate the effects of the cold compaction pressure as well as addition of TiH₂ and NH₄HCO₃ on the porous structure, phase components, superelastic behavior, and transformation temperature of the porous NiTi.

2. Experimental Procedures

Commercial elemental nickel (50-75 µm), titanium (50-75 μm), TiH₂ (15-40 μm), and NH₄HCO₃ (20-100 μm) powders, all with purity of 99.5%, were used in our study. Three different types of samples were fabricated by CF-HIP from different green compacts in order to assess the effects of the cold compaction pressure, TiH₂, and NH₄HCO₃ on the porosity, pore size, shape, and distribution. Samples of type A were fabricated from fully mixed equiatomic Ni and Ti powders compacted at different cold pressures, and samples of type B were prepared from fully mixed Ni, Ti and 9.0 wt.% TiH₂ powders with an atomic ratio of 1:1 of Ni to Ti at a cold compaction pressure of 200 MPa. Samples of type C were made from fully mixed equiatomic Ni and Ti powders with 30 wt.% NH₄HCO₃ at the same pressure. The materials were mixed for 12 h using a horizontal universal ball mill with a speed of 100 rpm and then pressed into green compacts in a steel mold with a diameter of 16 mm using a hydraulic press. For type C, the corresponding green compacts were preheated at 200 °C for 2 h in a tube furnace under a continuous flow of pure argon to remove $\rm NH_4HCO_3$. The green compacts were sintered in a HIP. Before sintering, the HIP chamber was evacuated and then backfilled with 99.995% pure argon. At first, the argon pressure was raised to 100 MPa at room temperature and then increased as the temperature increased. The argon pressure and temperature reached 150 MPa and 1050 °C at the same time. The high pressure and temperature were kept for 3 h allowing sufficient diffusion of nickel and titanium.

The phases and microstructures of the porous NiTi samples were determined by x-ray diffraction (XRD, Siemens D500), optical microscopy (OLYMPUS BH2-UMA) and stereo optical microscopy (OLYMPUS SZX12). The general porosity of the samples was evaluated by the equation: $\varepsilon = \left(1 - \frac{m}{\rho V}\right) \times 100(\%)$, where *m* and *V* are the mass and volume of porous samples, respectively, and ρ is the theoretical density of NiTi of 6.45 g/cm³ for bulk equiatomic NiTi. The samples for compression tests were machined into cylindrical bars 6 mm in diameter and 15 mm in length. The uniaxial compression tests were conducted at a constant displacement rate of 0.06 mm/min under room temperature of about 22 °C on an Instron 4206 to investigate the compression strength and SE. The transformation temperature was determined using a differential scanning calorimetry (DSC) with a heating/cooling rate of 5 °C/min.



Fig. 1 Porosities of porous NiTi and corresponding green compacts: (a) type A, (b) comparison of types A, B, and C

3. Results and Discussion

Figure 1(a) shows the porosities of the type A samples and corresponding green porosities at different cold compaction pressures. The final porosity of the porous NiTi falls in the range of 20-28 vol.%, indicating that the porosity is adjustable by changing the cold compaction pressure of the green compacts because the cold compaction pressure can alter the number, size and distribution of the original pores in the green compacts. Generally, the green porosity decreases with higher cold compaction pressure. However, the porous NiTi with the highest final porosity of about 28% was not sintered from the green compact at 50 MPa but from the one at 200 MPa. It suggests that the porosity of the porous NiTi is not determined only by cold compaction and the subsequent hot pressure during CF-HIP also significantly influences the final porosity. In this study, the hot pressure was 150 MPa, much higher than the cold compaction pressure of 50 MPa, thereby inducing larger deformation of this sample during sintering at 1050 °C. In comparison, with regard to the green compacts at 200 MPa, the hot pressure is close to the cold compaction pressure, and deformation is small during sintering. It is obvious that the final porosities of the samples are much lower than the corresponding green porosities (shown in Fig. 1a). This large drop is possibly due to the shrinkage of the original pores during CF-HIP. Figure 1(b) exhibits the difference in the porosity of the three types of samples and corresponding green compacts at a cold compaction pressure of 200 MPa. The final porosity of the type B samples rises to about 42%. It is due to the formation of a large amount of H2 pores from the decomposition of TiH2 during CF-HIP process (Ref 15). These H₂ pores favor the formation of open pores, and the quantity of closed pores decreases. In comparison, the final porosity of the type C



Fig. 2 Stereo optical macrographs of porous samples: (a) type A; (b) type B; and (c) type C



Fig. 3 Optical photographs of as-sintered porous NiTi: (a) type A; (b) type B; (c) stereo optical micrograph of type B; (d) high-magnification micrograph of (b); (e) type C; and (f) stereo optical micrograph of type C

samples increases to about 48% because its corresponding green porosity after removing NH_4HCO_3 is about 66% that is much higher than that of the type B samples. Figure 2(a)-(c) shows the optical macrographs of the as-sintered porous NiTi type A, type B. and type C samples produced from the corresponding green compacts at a cold compaction of 200 MPa, respectively. It is obvious that both TiH₂ and NH_4HCO_3 can further improve the final porosity of the as-sintered samples, while the latter can effectively change the pore size.

Figure 3 shows the microstructures of the as-sintered type A samples with 28% porosity as well as the type B and type

C samples made from the corresponding green compacts at a cold compaction of 200 MPa. It can be observed that the pores in the porous NiTi sintered from green compacts without TiH₂ are near circular and separate (shown in Fig. 3a). The open porosity of this sample is about 20% as measured by the ASTM B328-96 protocol (Ref 21) indicating that most of the pores are closed. It is believed that these closed pores are derived from the original pores in the green compacts. The formation of the other 20% open pores is possibly due to the penetration of high pressure argon gas during CF-HIP. In comparison, the open porosity of H₂ can

significantly improve the interconnection between separate pores (shown in Fig. 3b, c). Figure 3(d) illustrates the formation of the interconnected pores. In addition to the penetration of high pressure argon, high pressure H₂ can easily pierce the wall between separate pores in a soft medium at a high sintering temperature (indicated by the arrows in Fig. 3d). Figure 3(a) and (b) reveals that the sizes of most of the pores range from 20 to 200 µm, suggesting that TiH₂ has little effects on the pore size of the porous NiTi. It can be found in Fig. 3(e) and (f) that most of the pores in the type C samples are near circular have sizes ranging from 50 to 800 µm. The results suggest that besides the porosity, the space holder NH₄HCO₃ can also significantly improve the pore size but does not affect the pore shape compared with the type A samples. This is because NH₄HCO₃ can increase the number and size of the original pores in the green compacts upon removal. Hence, there are more Ar-filled pores with larger sizes in the type C samples resulting in the increase of the porosity and pore size after sintering. Furthermore, the addition of NH4HCO3 also sharply increases the open porosity of the as-sintered porous samples. The open porosity of the type C samples produced by the same protocol (Ref 21) is about 70% indicating that most of the pores are interconnected.

Figure 4 shows the phase compositions of the porous NiTi produced by CF-HIP. Several intermetallic compounds composed mainly of the intended B2 (austenite NiTi) and B19' (monoclinic martensite NiTi) phases together with minor second phases such as Ni₃Ti, Ti₂Ni, and Ni₄Ti₃. NiTi constitutes the major phase because of the equiatomic Ni-Ti composition of the green compacts before sintering according to the Ni-Ti binary diagram (Ref 22). The appearance of the second phases is a common phenomenon in the powder metallurgical processes of NiTi alloy (Ref 12, 14, 23), and it can be attributed to the composition fluctuation caused by the insufficient mixing of the raw powders and small particle size of the reagent powders (Ref 24). It is very difficult to completely remove some of the secondary phases such as Ni3Ti by solid-state diffusion because it is thermodynamically more stable than TiNi (Ref 23). It can be concluded from Fig. 4 that additions of TH₂ and NH₄HCO₃ do not influence the phase components of the porous NiTi.



Fig. 4 Phase components of porous NiTi types A, B, and C. Δ : B2 (NiTi); ∇ : B19' (NiTi); \blacksquare : Ni₃Ti; \bigcirc : Ti₂Ni; \square : Ni₄Ti₃

The compressive stress-strain results of the porous NiTi are shown in Fig. 5. The compressive strength of the type A samples is about 270 MPa when strained up to 3.5%, and the residual strain is about 0.05% after unloading, whereas the corresponding strength and residual strain of the type B samples are 220 MPa and 0.1%, respectively. In addition, curve 1 exhibits an obvious stress-plateau, but curve 2 does not show it clearly. The data suggest that TiH₂ addition to the green compacts slightly impairs the compression strength and superelasticity of the porous NiTi. This is because TiH₂ addition not only improves the porosity but also interconnects the separate closed pores. In comparison with type A and type C samples, the irregular pore shape in type B samples (shown in Fig. 3) would be harmful to the mechanical properties of porous NiTi alloys. The corresponding strength and residual strain of the type C samples are about 130 MPa and 0.12%, respectively, when strained up to 2.8%. It is well accepted that the similar Young's modulus with human bone favors the smooth transfer of a load and thus is beneficial to the formation of new bone tissues on the implants (Ref 4, 8). As is known, some hard tissues like bone and tendon possess superelastic behavior, i.e., their stress-strain curves can exhibit stress-plateau despite the extension of stain. Therefore, the mechanical properties of both type A and type C samples are more similar to natural bone than those of type C samples. The Young's moduli of the porous samples can be calculated in the near-linear region by Hooke's Law, i.e., $E = \sigma/\epsilon$, where E represents the elastic modulus, and σ is the stress when the strain value reaches to ε . According to the stress-strain curves in Fig. 5, the Young's moduli of type A, type B, and type C samples are 8.648, 5.06, and 6.445 GPa, respectively, which are very close to the value of natural bone. According to the reported results (Ref 25, 26), the Young's modulus of cancellous bones is in the range of 1-2 GPa, while the value of wet compact human bones varies from 11.5 to 17.0 GPa. Compared with porous NiTi fabricated by other methods (Ref 12, 14, 16), the porous NiTi prepared by CF-HIP exhibits higher compression strength, high strain recovery, and narrower hysteretic loop because during the CF-HIP process, the high pressure argon gas prevents densification and enhances bridging among particles. The evolution of the compressed argon gas inside the compact helps in eliminating the microcracks. Although Zhao et al. (Ref 27) have reported that porous NiTi with 13% porosity fabricated by



Fig. 5 Compressive stress-strain curves of porous NiTi type A, B and C



Fig. 6 DSC curves of porous NiTi annealed at 450 °C for 1 h

spark plasma sintering (SPS) shows higher compression strength and excellent SE similar to those of dense NiTi, the compression strength of their 25% porosity sample is much lower than that of our type C sample with 48% porosity. This also confirms that porous NiTi fabricated by CF-HIP can have good mechanical properties in conjunction with higher porosity.

Figure 6 shows the DSC curves of the porous NiTi annealed at 450 °C for 1 h. There are two transformation peaks in the cooling curves, namely, B2 \rightarrow R and R \rightarrow B19', as well as one broadened B19' \rightarrow B2 transformation peak in the reverse direction. Comparing the three types of samples, it is evident that the transformation temperatures shift to lower values with increased porosities. This is possibly due to the non-uniform internal stress caused by pore's expansion. The internal stress increases with higher porosities and suppresses martensite transformation. Figure 6 shows that the transformation temperatures of the samples of types A, B, and C are favorable from the viewpoint of the human body's temperature.

4. Conclusions

In summary, porous NiTi has been successfully fabricated by CF-HIP. The porosity is adjustable by changing the cold compaction pressure of the green compacts, TiH₂, or NH₄HCO₃ addition. Compared with materials synthesized by other methods, the porous NiTi produced by CF-HIP exhibits higher compression strength and excellent SE in addition to the higher porosity. The transformation temperatures of the three types of porous NiTi favor superelasticity at the human body temperature thereby boding well for the use of the materials in orthopedic corrections.

Acknowledgments

This study was jointly supported by City University of Hong Kong Applied Research Grant (ARG) No. 9667038, National Natural Science Foundation of China No. 50901032, 51101053, Ministry of Education Specialized Research Foundation for Doctoral Program of Universities No. 20094208120003, Hubei Provincial Middle-Young Research Fund Grant no. Q20101010, and Wuhan ChenGuang Research Program Grant no. 201150431134.

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